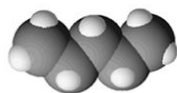
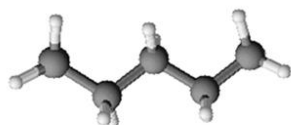
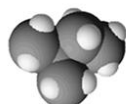
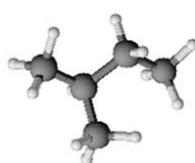


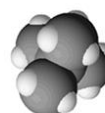
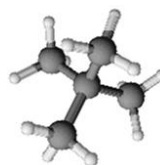
1. Il y a 3 isomères du pentane.



n-pentane



méthylbutane



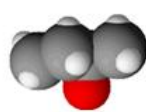
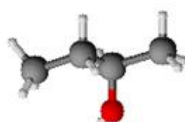
diméthylpropane

2.

a. Il y a 3 alcools isomères de formule brute $C_4H_{10}O$.



butan-1-ol



butan-2-ol

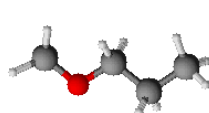


2-méthylpropan-1-ol

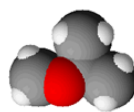
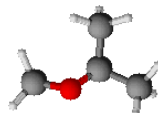


2-méthylpropan-2-ol

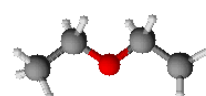
b. Et 3 Etheroxydes de formule brute $C_4H_{10}O$.



1-méthoxypropane



2-méthoxypropane



éthoxyéthane

Les alcools et les étheroxydes sont des isomères de nature de fonction.